Cruise: SAV1803  
Ship: R/V Savannah  
Dates: March 8th, 2018 – March 13th, 2018  
Expo Code: 33H620180308  
Chief Scientist: Dr. Chris Kelble  
Equipment: CTD  
Total number of stations: 7  
Location: Southwest Florida Gulf of Mexico coastal region

Samples were collected for Dr. Leticia Barbero for the Ocean Acidification Program during the South Florida Project (SFP) water quality cruises in the SW Gulf of Mexico lead by Dr. Chris Kelble.

**Sample Collection**

The discrete samples were collected from Niskin bottles attached to a 12 bottle configured rosette onboard the R/V Savannah. The samples were collected by Dr. Chris Kelble, Ian Smith and Kelly Montenero. The date and time listed in the data file are UTC when each sample bottle was collected.

**DIC:**
7 locations, 7 samples each 500-ml, 0 duplicate samples.  
Sample_ID#: 101, etc.; Bottle number  
PI: Dr. Rik Wanninkhof  
Analyzed by: Charles Featherstone and Patrick Mears

**pH:**
7 locations, 7 samples each 500-ml, 0 duplicate samples.  
Sample_ID#: 101, etc.; Bottle number  
PI: Dr. Rik Wanninkhof  
Analyzed by: Charles Featherstone and Patrick Mears

**TAlk:**
7 locations, 7 samples each 500-ml, 0 duplicate samples.  
Sample_ID#: 101, etc.; Bottle number  
PI: Dr. Rik Wanninkhof and Leticia Barbero  
Analyzed by: Charles Featherstone and Patrick Mears

**Sample Analysis**

**DIC:**

<table>
<thead>
<tr>
<th>Instrument ID</th>
<th>Date</th>
<th>Certified CRM (µmol/kg)</th>
<th>CRM Value (µmol/kg)</th>
<th>CRM Offset (µmol/kg)</th>
<th>Blank (Counts)</th>
<th>Avg. Sample Analysis Time</th>
</tr>
</thead>
<tbody>
<tr>
<td>AOML 5</td>
<td>11/29/2017</td>
<td>2042.41</td>
<td>2043.27</td>
<td>0.86</td>
<td>20.0</td>
<td>7.7</td>
</tr>
</tbody>
</table>
Analysis date: 08/10/2018  
Coulometer used: DICE-CM5011- AOML 5  
Blanks: 20.0 counts/min  
CRM # 665 was used and with an assigned value of (includes both DIC and salinity):  
Batch 173, c: 2042.41 µmol/kg, S: 33.414  
CRM values measured: AOML 5: offset 0.86 µmol/kg (2043.27 µmol/kg).  
Average run time, minimum run time, maximum run time: 7.7, 7 and 9 min.  

**Reproducibility:**  
No duplicate samples were taken

CRM, salinity and HgCl2 correction applied: Salinity correction was applied using TSG salinity.

**Remarks**  
The volume correction was applied due to added HgCl2 (Measured DIC*1.00037).  
The first CRM of each cell was used for a CRM correction.

The DIC instruments were stable: the gas loop and CRM values did not change significantly throughout the life span of each cell.

The blank on AOML 5 (08/10/2018) was raised from 12.0 to 20.0 before running the CRM.

The samples were analyzed using the DICE (AOML 5) and a coulometer from UIC, Inc. CM5011 with CM5011 emulation software.

**pH:**  
Analysis date: 08/10/2018  
Spectrophotometer used: HP Agilent 8453  

A CRM #536 from Batch 173 was run before analysis of samples; pH value = 7.8769

**Reproducibility:**  
No duplicate samples were taken

<table>
<thead>
<tr>
<th>Sample ID</th>
<th>Station</th>
<th>Bottle #</th>
<th>Salinity</th>
<th>Temperature</th>
</tr>
</thead>
<tbody>
<tr>
<td>CRM173-536</td>
<td>CRM</td>
<td>536</td>
<td>33.414</td>
<td>20.038</td>
</tr>
<tr>
<td>54</td>
<td>54</td>
<td>103</td>
<td>32.113</td>
<td>20.041</td>
</tr>
</tbody>
</table>
The equations of Liu et al, 2011 formulated using the purified m-cresol purple indicator was used to determine pH of the samples. pH samples were analyzed at 20°C at Full Scale (pH 0-14).

Samples were run on an automated system where the temperature was kept constant.

Approximately 80 mL of sample was extracted from each DIC sample bottle by syringe before DIC analysis to determine the pH.

Samples are reported at 25°C.

**TA1k:**

Analysis date: 8/20/2018  
Titration system used: Open cell  
CRM Batch 173, Salinity = 33.414, cert. TA = 2210.77 µmol/kg.

On 8/20/2018 one CRM was analyzed before the samples and the same CRM was run at the end of analysis each day for each system.

The TA for the water samples was corrected using the first CRM (bottle # 689) ratio between the certified and measured values of the CRMs run on each cell. The following table shows the CRM measurements for each day and cell.

| Cell System | Date       | Time       | Bottle # | TA    | |ΔCRM| |
|-------------|------------|------------|----------|-------|-----|-----|
| 2           | 8/20/2018  | 08:36:53   | 689      | 2216.60 | 5.31 |
| 2           | 8/20/2018  | 15:20:52   | 575      | 2221.92 | 5.31 |

**Reproducibility:**  
No duplicate samples were taken

**Remarks**

The CRM measurement for each day was used to correct the data for that day only.
Comments

The latitude, longitude, date, and time reported with the DIC, pH and TA\textsubscript{alk} measurements were taken from the sample field log. The field log values are provided for reference; no post-cruise assurance of accuracy has been done to this data. The Niskin bottles are approximately one half meter above the CTD sensors on the rosette. Therefore, Temp and Sal are bin-averaged CTD values representing the next shallower depth from that recorded by the CTD (CTD Depth) at the time the Niskin bottles were fired with the exception of the surface values, which are the same as the CTD Depth values (as per the log sheet).

The Sample ID is the sample station, cast number and Niskin bottle number for the discrete samples.

Nutrients:

Analysis Date: April 4\textsuperscript{th} and 5\textsuperscript{th}, 2018

Nutrient samples were analyzed using a Seal Analytical high resolution digital colorimeter auto-analyzer 3 (AA3). A series of standards for each method were run before sample analysis to obtain a calibration curve for data reduction. Method 353.4 was used to determine the concentration of nitrate and nitrite for each station (Zhang et al., 1997b). This method used automated, gas-segmented, continuous flow colorimetry for the analysis of nitrate and nitrite. Samples were first passed through a copper-coated cadmium reduction column. Nitrate was reduced to nitrite in a buffer solution. The nitrite was then determined by diazotizing with sulfanilamide and coupling with N-1-naphthylethylenediamine dihydrochloride to form a color azo dye. The absorbance measured at 550 nm is linearly proportional to the concentration of nitrite + nitrate in the sample. Nitrate concentrations are obtained by subtracting nitrite values, which have been separately determined without the cadmium reduction procedure, from the nitrite + nitrate values.

Method 365.5 was used to determine the concentration of orthophosphate for each station (Zimmermann and Keefe, 1997; Zhang et al., 2001). This method used automated colorimetric and continuous flow analysis for the determination of low-level orthophosphate concentrations. Ammonium molybdate and antimony potassium tartrate react in an acidic medium with orthophosphate to form an antimony-phospho-molybdate complex. This complex was reduced to a blue-colored complex by ascorbic acid. The absorbance measured at 880 nm is proportional to the phosphate concentration in the sample.

Method 366.0 was used to determine the concentration of soluble silica for each station (Zhang and Berberian, 1997). This method used automated, gas-segmented, continuous flow colorimetry for the analysis of dissolved silicate concentration. In this method, $\beta$-molybdosilicic acid was formed by the reaction of the silicate contained in the sample with molybdate in acidic solution. The $\beta$-molybdosilicic acid was then reduced by ascorbic acid to form molybdenum blue. The absorbance of the molybdenum blue, measured at 550 nm, is linearly proportional to the concentration of silicate in the sample.


**UPDATE:**
Between March and June of 2021, all of the data for the discrete samples was put into a uniform format. The supporting information was checked for accuracy, especially the expocode, date, time, and positions.
Additionally, pH results were recalculated to 20 and 25 degrees Celsius.